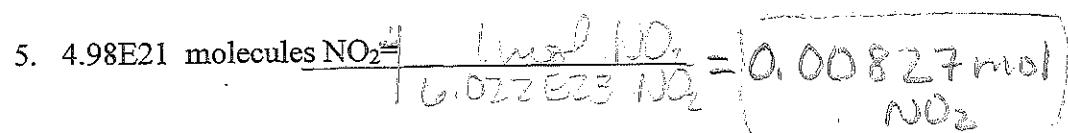
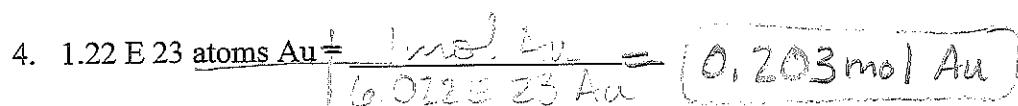
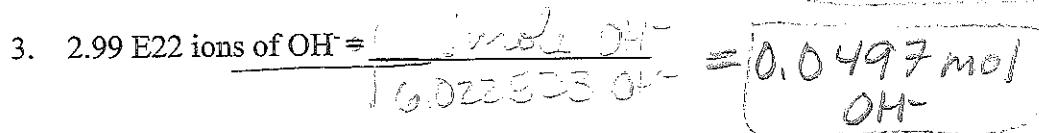
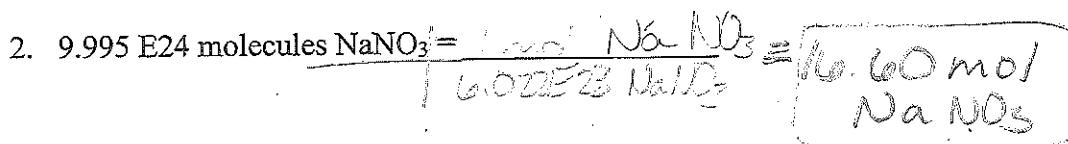
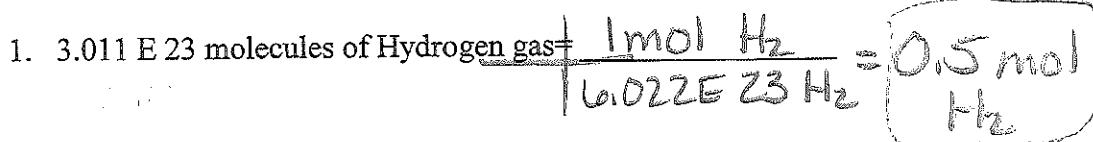


(#6-1)
 Chemistry
 Stoichiometry
 Mole #1

$$1 \text{ mole} = 6.022 \times 10^{23} \text{ things}$$

Convert the following to moles.



Convert the following from moles to number of particles

